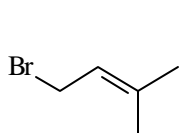
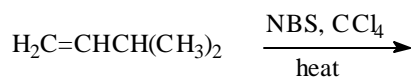
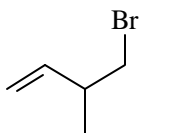


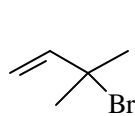
1. What is(are) the expected product(s) of the following reaction?



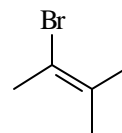
A



B

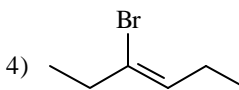
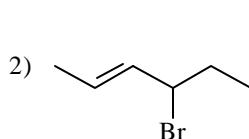
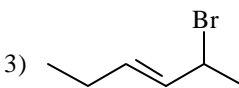
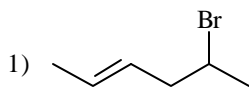
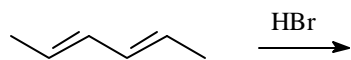


C

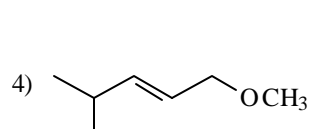
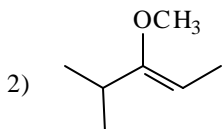
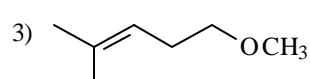
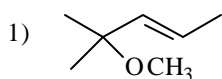
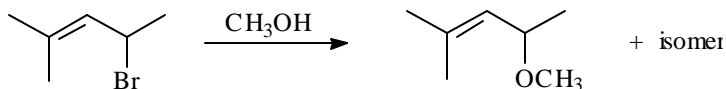


D

- 1) only B 2) only C 3) A and C 4) B and D
2. Which of the following is the 1,4-addition product in the reaction shown below?

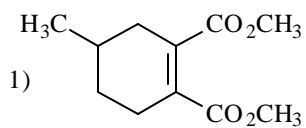
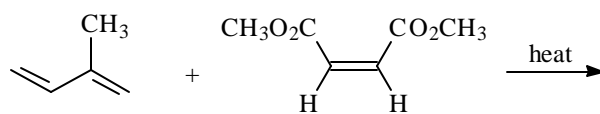


- 1) 1 2) 2 3) 3 4) 4
3. Methanolysis of 4-bromo-2-methyl-2-pentene gives two isomeric substitution products, one of which is shown. What is the other substitution product?

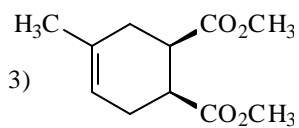


- 1) 1 2) 2 3) 3 4) 4

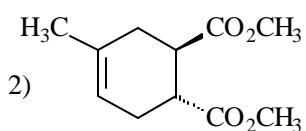
4. What is product of the following Diels-Alder reaction?



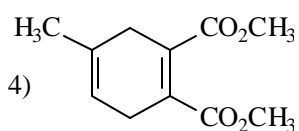
+ enantiomer



+ enantiomer



+ enantiomer



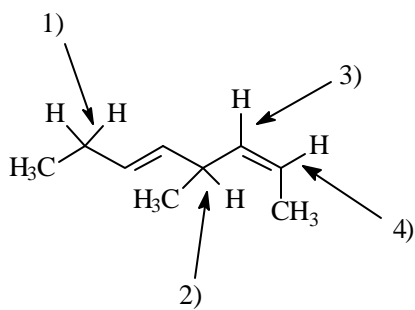
1) 1

2) 2

3) 3

4) 4

5. Identify the weakest carbon-hydrogen bond in the following diene?



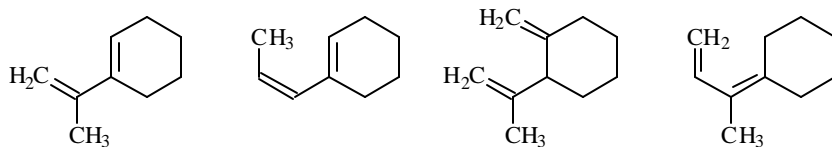
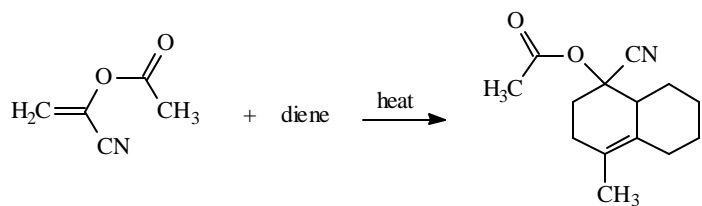
1) 1

2) 2

3) 3

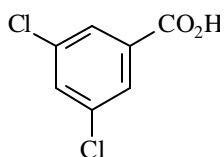
4) 4

6. Identify the diene used in the reaction shown below.



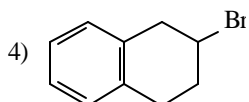
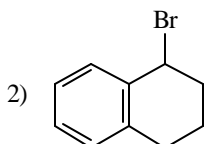
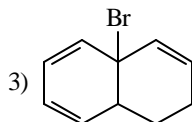
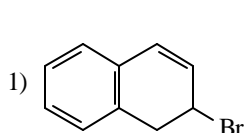
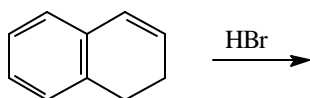
- 1) 1 2) 2 3) 3 4) 4

7. What is the IUPAC name of the following compound?



- 4,6-dichloro-2-benzoic acid
 - 2,4-dichlorobenzoic acid
 - 3,5-dichlorobenzoic acid
 - meta*-dichlorobenzoic acid
8. Identify the aromatic compounds.
-
- A and B
 - C and D
 - A, B, and D
 - all of them
9. Which of the following ions are aromatic species?
-
- A and C
 - B and C
 - B and D
 - C and D

14. Predict the major organic product in the following reaction.



1) 1

2) 2

3) 3

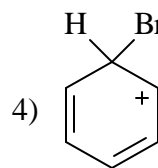
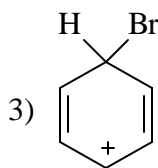
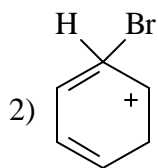
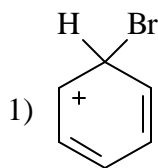
4) 4

15. In which of the following are carbon-carbon bond lengths arranged in the correct order?

shortest _____ longest

- | | | |
|----------------|-------------|-------------|
| 1) benzene | ethylene | cyclohexane |
| 2) ethylene | cyclohexane | benzene |
| 3) cyclohexane | benzene | ethylene |
| 4) ethylene | benzene | cyclohexane |

16. Which one of the following is not a resonance form of the cyclohexadienyl cation intermediate in the bromination of benzene?



1) 1

2) 2

3) 3

4) 4

17. Starting with toluene, which of the following is the best synthesis of *meta*-bromobenzoic acid?

1) (1) $\text{Br}_2, \text{FeBr}_3$ (2) $\text{K}_2\text{Cr}_2\text{O}_7, \text{H}_2\text{SO}_4$ (heat)

2) (1) $\text{Br}_2, \text{h}\nu$ (2) $\text{K}_2\text{Cr}_2\text{O}_7, \text{H}_2\text{SO}_4$ (heat)

3) (1) $\text{K}_2\text{Cr}_2\text{O}_7, \text{H}_2\text{SO}_4$ (heat) (2) $\text{Br}_2, \text{FeBr}_3$

4) (1) $\text{K}_2\text{Cr}_2\text{O}_7, \text{H}_2\text{SO}_4$ (heat) (2) $\text{Br}_2, \text{h}\nu$

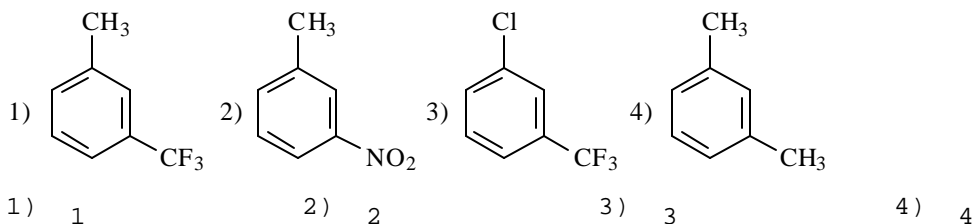
1) 1

2) 2

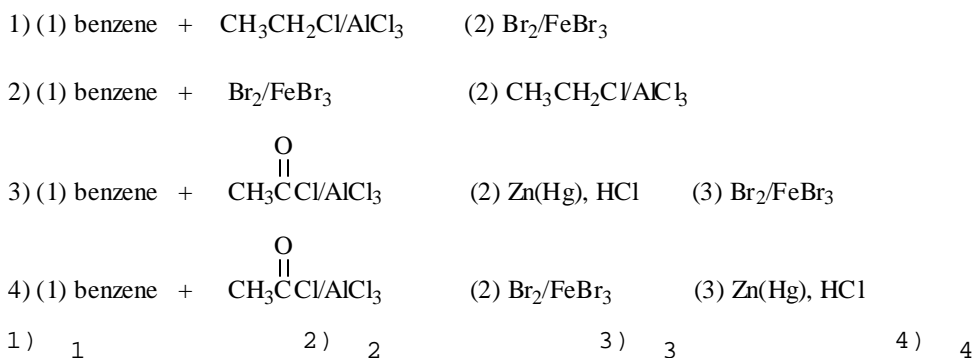
3) 3

4) 4

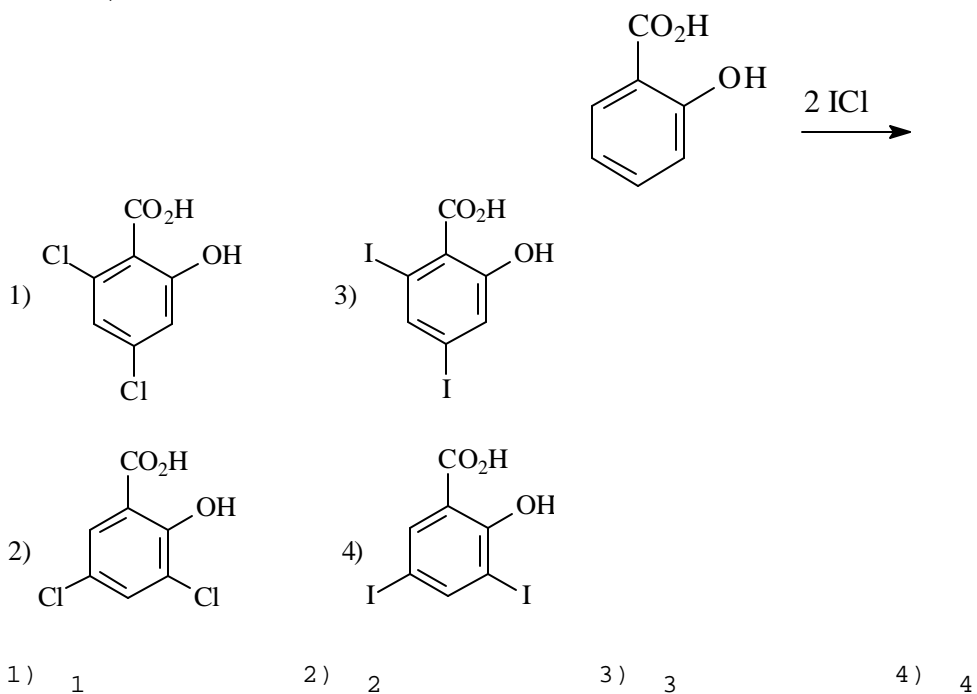
18. Which one of the following compounds undergoes electrophilic aromatic sulfonation at the fastest rate?



19. Which of the following is the best method to make *meta*-bromoethylbenzene from benzene?



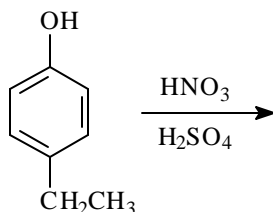
20. Salicylic acid reacts with two equivalents of ICl to give one of the products below. Which one is it? (hint: Cl is more electronegative than I)



21. Which isomer of dichlorobenzene gives a single mononitration product?

- 1) ortho
- 2) meta
- 3) para
- 4) none of them

22. What is the major product of the following reaction?



- | | |
|---------------------------|--------------------------|
| 1) 4-ethyl-2-nitrophenol | 2) 4-ethyl-3-nitrophenol |
| 3) 1-ethyl-4-nitrobenzene | 4) 4-nitrophenol |

23. Which of the following is the best method to make *n*-butylbenzene?

1) benzene + $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{Cl}/\text{AlCl}_3$

2) benzene + $\text{CH}_3\text{CH}_2\text{CH}=\text{CH}_2/\text{H}_2\text{SO}_4$

3) (1) benzene + $\text{CH}_3\text{CH}_2\text{CH}_2\overset{\text{O}}{\parallel}\text{C}/\text{AlCl}_3$ (2) $\text{H}_2\text{NNH}_2/\text{KOH}$, heat

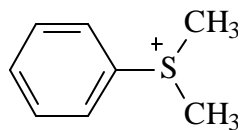
4) (1) benzene + $\text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_2\text{OH}/\text{H}_2\text{SO}_4$

1) 1 2) 2 3) 3 4) 4

24. In the Friedel-Crafts alkylation of benzene, dialkylation is often a significant byproduct. In the Friedel-Crafts acylation of benzene, diacylation is not a significant byproduct. Which of the following is the primary reason for this difference?

- 1) Alkyl groups activate the ring to further substitution, acyl groups deactivate it.
- 2) Alkyl groups are less sterically hindered than acyl groups.
- 3) Acyl cations are more difficult to make with Lewis acids.
- 4) Unlike acyl cations, carbocations can undergo rearrangements.

25. Predict the effect the substituent attached to the benzene ring below would have on electrophilic aromatic substitution reactions?



- 1) ortho/para director, activator
- 2) ortho/para director, deactivator
- 3) meta director, activator
- 4) meta director, deactivator